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(54) Title: USE OF GENTISIC ACID OR GENTISYL ALCOHOL FOR STABILISING RADIOLABELED PEPTIDES AND PROTEINS

(57) Abstract

Gentisic acid and its derivatives substantially inhibit peptide autoradiolysis. Gentisic acid or its derivatives may also be used in combination with other stabilizers such as inositol, ascorbic acid, and citrate to inhibit autoradiolysis of radiolabeled peptides.

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USE OF GENTISIC ACID OR GENTISYL ALCOHOL FOR STABILISING RADIOLABELED PEPTIDES AND PROTEINS

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BACKGROUND OF THE INVENTION

This invention relates to stabilizers for radiopharmaceutical compositions. More particularly, stabilizers such as gentisic acid and its derivatives, alone or in combination with other stabilizers, are used to inhibit autoradiolysis of radiolabeled peptides and proteins.

The number of therapeutic and diagnostic uses of radiolabeled compositions is continually growing. Such uses generally involve the introduction of a suitable radiolabeled composition into a biological subject. Detection and imaging of radioactive emissions may be used to diagnose and locate the presence of aberrations, pathological conditions, and the like. In some cases, the radiolabeled composition may be designed to locate in or to seek out specific tissues or biological receptors for the purpose of delivering therapeutic radioactive emissions.

In general, a radiolabeled composition comprises a radionuclide, a carrier agent designed to target the specific organ of interest, various auxiliary agents which affix the radionuclide to the carrier, a delivery vehicle, such as water, suitable for injection into or aspiration by the patient, physiologic buffers and salts, and the like.

Some radiopharmaceutical preparations are known to For example, technetium-99m and require stabilizers. rhenium-186 compositions are unstable in oxygen and require stabilizers, such as antioxidants or reducing agents, to maintain the technetium or rhenium in a usable oxidation state. Typical reducing agents used in technetium-99m and rhenium-186 compositions include stannous, ferrous, and Sometimes other additives, such as chromous salts. ascorbic acid, d-ascorbic acid, gentisic acid, reductic acid, p-aminobenzoic acid, erythorbic hydroxybenzoic acid, nicotinic acid, nicotinamide, and 2,510

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dihydroxy-1,4-benzenedisulfonic acid, are included to inhibit the oxidation of the radionuclide or the reducing agent.

Other radionuclides, such as $^{\rm HI}$ In, 90 Y, and 67 Ga exist in a stable oxidation state, and therefore, do not require stabilizers to maintain their useful oxidation state.

Over the years, there has been growing interest in radiolabeled proteins such as hormones, preparing macroaggregated albumin ("MAA"), human serum albumin ("HSA"), monoclonal antibodies, or monoclonal antibody fragments for the purpose of diagnosing and treating diseases, such as inflammation, deep vein thrombosis, or In some cases, autoradiolysis of the labeled cancer. protein has been observed. To inhibit or prevent autoradiolysis, experts have suggested adding HSA to the composition (e.g., R.A.J. Kishore, et al., "Autoradiolysis of Iodinated Monoclonal Antibody Preparations," Int. J. Radiat. Appl. Instrum., Part B, Vol. 13, No. 4, pp. 457-459 (1986)) or keeping the radiopharmaceutical composition frozen between preparation and administration (e.g., R.L. Wahl, et al., "Inhibition of Autoradiolysis of Radiolabeled Monoclonal Antibodies by Cryopreservation, " J. Nuc. Med., Vol. 31, No. 1, pp. 84-89 (1990)). These techniques for preventing autoradiolysis are often not effective or practical when used with many radiolabeled peptides and proteins.

Recently, a number of exciting new peptides for diagnostic and therapeutic applications have been isolated and synthetically developed. One such peptide is an octapeptide somatostatin analog known as octreotide and described in U.S. Patent No. 4,395,403. Octreotide has a very high binding affinity to somatostatin receptors in a variety of human tumors. By linking octreotide to a suitable chelating agent capable of forming a complex with radionuclides, it has been possible to create radiolabeled

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octreotide which effectively images tumors having somatostatin receptors. Somatostatin analogs containing chelating groups are described in UK Patent Publication No. 2,225,579.

Despite the potential usefulness of radiolabeled peptides, it has been found that they are very susceptible to autoradiolysis. As used herein, the term autoradiolysis includes chemical decomposition of the peptide by the action of radiation emitting from the radioisotope coupled to the peptide. Some believe autoradiolysis may be caused 10 by the formation of free radicals, such as hydroxyl radicals, in the water or delivery vehicle by the radiation emitted from the radioisotope.

From the foregoing, it will be appreciated that what is needed in the art are stable radiolabeled peptide and protein compositions. Thus, it would be a significant advancement in the art to provide stabilizing agents which substantially inhibit autoradiolysis of radiolabeled peptides and proteins.

Such compositions for substantially inhibiting peptide and protein autoradiolysis are disclosed and claimed herein.

BRIEF SUMMARY OF THE INVENTION

The present invention provides compositions for 25 preparing stable radiolabeled peptide and protein The stabilizers used in the present preparations. invention are able to substantially inhibit autoradiolysis of radiolabeled peptides and proteins. It has been found that stabilizers that are effective at preventing oxidation of radioisotopes, such as 99mTc, are not necessarily effective at preventing autoradiolysis of peptides and proteins. Likewise, it has been found that stabilizers and techniques used to prevent autoradiolysis of proteins, such as addition of HSA or freezing, are not effective or 35

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practical in many cases. Accordingly, the present invention is directed to compositions containing stabilizers that substantially inhibit autoradiolysis of peptides and proteins.

Gentisic acid and its derivatives have been found to be very effective at inhibiting peptide and protein autoradiolysis. Gentisic acid or its derivatives may also be used in combination with other stabilizers, such as inositol and ascorbic acid, to inhibit autoradiolysis.

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DETAILED DESCRIPTION OF THE INVENTION

The present invention is directed to compositions inhibit substantially that stabilizers containing autoradiolysis of peptides and proteins. One class of very effective at inhibiting is stabilizers that autoradiolysis of peptides and proteins is gentisic acid The radiolabeled peptides and and its derivatives. proteins for which the stabilizers of the present invention are needed include peptides and proteins having diagnostic and therapeutic applications.

Gentisic acid (chemically: 2,5-dihydroxybenzoic acid) is commercially available and can be prepared by several methods known in the art. Derivatives of gentisic acid include the pharmaceutically-acceptable salts and esters of gentisic acid and gentisyl alcohol. Examples of suitable gentisic acid derivatives are described in U.S. Patent Nos. 4,497,744 and 4,232,000 which are incorporated herein by reference.

Other stabilizers such as inositol and ascorbic acid may be used in combination with gentisic acid or its derivatives to inhibit autoradiolysis of radiolabeled peptides.

The pharmaceutically-acceptable salts and esters of gentisic acid and gentisic alcohol can be prepared by standard neutralization and esterification procedures known

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in the art, such as the techniques described in U.S. Patent Nos. 4,497,744 and 4,232,000. In practice, the salts and esters of gentisic acid suitable for use in the present invention can be selected according to their solubility. Soluble gentisate salts include the soluble alkali metal, alkaline earth metal, heavy metal, and ammonium salts. The alkali metal salts, such as sodium, lithium, and potassium, are very soluble and are currently preferred. The alkaline earth metal gentisate salts, such as calcium and magnesium, are less soluble, but are still suitable for use herein.

The present invention may be used with a wide range of radioisotopes capable of causing autoradiolysis of peptides. Such radioisotopes include γ -, β -, and α -emitters.

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Suitable γ -emitters include those radionuclides which are useful for diagnostic techniques. Examples of some typical γ -emitting radionuclides are 67 Ga, 111 In, 99m Tc, 169 Yb, and 125 I, 123 I, and 201 Tl. Examples of possible β -emitting radionuclides include those which are useful in therapeutic applications such as 90 Y, 67 Cu, 186 Re, 188 Re, 169 Er, 121 Sn, 127 Te, 143 Pr, 198 Au, 109 Pd, 165 Dy, 32 P, 142 Pr, 177 Lu, 166 Ho, 153 Sm, 90 Y, 131 I, 89 Sr, and 105 Rh. Typical α -emitters include 212 Bi, 211 At, 241 Am, and 255 Fm.

Radiolabeling of peptides and proteins can be achieved using various methods known in the art. For example, peptides can be labeled through use of a bifunctional chelate, direct labeling, or covalent binding to a specific functional group of an amino acid side chain. The use of a bifunctional chelate involves covalent attachment of a chelate, which complexes with the radionuclide, to the peptide or protein. Possible bifunctional chelates include DTPA and N₃S ligands. The DTPA may be attached to the peptide or protein by the dicyclic dianhydride method described in U.S. Patent No. 4,479,930, which is incorporated herein by reference. N₃S ligands may be

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attached to the peptide or protein by the methods described in U.S. Patent No. 4,965,392 and European Patent Publication Number 0284071, which are incorporated herein by reference.

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In direct labeling, the radionuclide binds to the functional group of amino acid side chains present in the peptide or protein. The radionuclide may also bind to reduced forms of a peptide or protein, such as a peptide or protein containing a reduced disulfide bond. One example of direct labeling known in the art is described in U.S. Patent No. 4,877,868, which is incorporated by reference.

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Another well known technique for labeling peptides and proteins involves covalently binding the radionuclide to one specific functional group of an amino acid side chain, such as incorporation of iodide into the phenol group of a tyrosine residue.

Commercial products for preparing radiopharmaceuticals are generally provided as lyophilized (freeze-dried) "kits" or as liquid formulations. Lyophilized kits are well known in the art. According to the present invention, lyophilized kits may contain a transfer ligand, such as citric acid, acetic acid, or sodium tartrate, a reducing agent, depending on the radioisotope that is used, a bulking agent, such as inositol or lactose, the peptide or protein to be labeled, and one or more gentisic acid stabilizers may also Additional stabilizers. incorporated into the formulation as described herein. The radioisotope is added to the lyophilized kit just prior to patient administration.

Liquid formulations usually contain the peptide or protein labeled with the radioisotope. According to the present invention, the liquid formulation also contains one or more gentisic acid derivatives to stabilize the formulation. Other stabilizers, such as inositol and ascorbic acid, may also be included in the formulation to

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improve stability. A surfactant, such as polysorbate 80, and a salt solution to give a desired ionic strength, may also be added to improve stability if the solution is autoclaved. In some cases, the solution may need to be deoxygenated.

The radiolabeled compositions of the invention may be administered parenterally, preferably intravenously, in the form of injectable pharmaceutical solutions or suspensions according to conventional techniques.

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Dosages employed in practicing the therapeutic method of the present invention will of course vary depending on the particular condition to be treated, for example the volume of the tumor, the particular chelate employed, the half-life of the radioisotope, and the therapy desired. In general, the dose is calculated on the basis of radioactivity distribution to each organ and on observed target uptake.

The following examples are offered to further illustrate different aspects of the present invention. These examples are intended to be purely exemplary and should not be viewed as a limitation on any claimed embodiment.

Example 1

Preparation of "In labeled DTPA-octreotide with citrate additive.

To a lyophilized kit containing 10 μ g DTPA-octreotide (N-[3,6,9,9-tetrakis(carboxymethyl)-3,6,9-triazanonanoyl]-D-phenylalanyl-L-hemicystyl-L-phenylalanyl-D-tryptophyl-L-lysyl-L-threonyl-L-hemicystyl-L-threoninol cyclic (2-7) disulfide, MW=1394.60 gm/mol), 5.6 mg trisodium citrate dihydrate, and 0.4 mg citric acid monohydrate was added 1.0 ml of 0.02 M HCl containing 4.70 mCi of ¹¹¹In. The pH of the solution was 4.5. The solution was kept at room temperature and monitored for the amount of ¹¹¹In DTPA-octreotide using reverse-phase HPLC and a Beckman 170

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radiometric detector. The HPLC method used a Hamilton PRP-1 column, 25 cm x 4.1 mm, 10 microns and a gradient system, linearly ramping from 100% A (A=10:90 ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) to 43:57 A:B (B=50:50 ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) over 10 minutes, then held at 43:57 A:B for 15 minutes, then linearly ramped to 100% B over 10 minutes. The flow rate was 1.5 ml/min. The retention time of ¹¹¹In DTPA-octreotide was 30 minutes.

The purity of the ^{III}In labeled peptide was 87% immediately post-reconstitution and 50% at 15 hours post-reconstitution.

Example 2

<u>Preparation of "IIIn labeled DTPA-octreotide</u> with citrate and inositol additives.

To a lyophilized kit containing 10 μ g DTPA-octreotide (N-[3,6,9,9-tetrakis(carboxymethyl)-3,6,9-triazanonanoyl]-D-phenylalanyl-L-hemicystyl-L-phenylalanyl-D-tryptophyl-Llysyl-L-threonyl-L-hemicystyl-L-threoninol cyclic (2→7) disulfide, MW=1394.60 gm/mol), 5.6 mg trisodium citrate dihydrate, 0.4 mg citric acid monohydrate, and 1.0 mg inositol was added 1.0 ml of 0.02 M HCl containing 6.04 mCi of 111 In. The pH of the solution was 4.5. The solution was kept at room temperature and monitored for the amount of $^{
m III}$ In DTPA-octreotide using reverse-phase HPLC and a Beckman 170 radiometric detector. The HPLC method used a Hamilton PRP-1 column, 25 cm x 4.1 mm, 10 microns and a gradient from ramping 100% linearly system, ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) to 100% B (B=50:50 ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) over 60 minutes. The flow rate was 1.5 ml/min. The retention time of "In DTPA-octreotide was 44-45 minutes.

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The purity of the "IIIIn labeled peptide was 94% immediately post-reconstitution and 73.5% at 24 hours post-reconstitution.

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Example 3

Preparation of "In labeled DTPA-octreotide with citrate, inositol, and ascorbic acid additives.

To a lyophilized kit containing 10 μg DTPA-octreotide (N-[3,6,9,9-tetrakis(carboxymethyl)-3,6,9-triazanonanoyl]-D-phenylalanyl-L-hemicystyl-L-phenylalanyl-D-tryptophyl-L-10 lysyl-L-threonyl-L-hemicystyl-L-threoninol cyclic (2→7) disulfide, MW=1394.60 gm/mol), 5.6 mg trisodium citrate dihydrate, and 0.4 mg citric acid monohydrate was added 1.0 mg inositol, 8.8 mg ascorbic acid, 9.9 mg sodium ascorbate, and 1.0 ml of 0.02 M HCl containing 5.05 mCi of "IIIIn. 15 pH of the solution was 4.0. The solution was kept at room temperature and monitored for the amount of "In DTPAoctreotide using reverse-phase HPLC and a Beckman 170 radiometric detector. The HPLC method used a Hamilton PRP-1 column, 25 cm x 4.1 mm, 10 microns and a gradient 20 100% ramping from (A=10:90 linearly ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) to 100% B (B=50:50 ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) over 35 minutes. The flow rate was 1.5 ml/min. The retention time of "III DTPA-octreotide was 30" 25 minutes.

The purity of the ^{III}In labeled peptide was 94% immediately post-reconstitution and 86% at 22 hours post-reconstitution.

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Example 4

Preparation of "In labeled DTPA-octreotide with citrate, inositol, and gentisic acid additives.

To a lyophilized kit containing 10 µg DTPA-octreotide

(N-[3,6,9,9-tetrakis(carboxymethyl)-3,6,9-triazanonanoyl]
D-phenylalanyl-L-hemicystyl-L-phenylalanyl-D-tryptophyl-L-

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lysyl-L-threonyl-L-hemicystyl-L-threoninol cyclic (2-7) disulfide, MW=1394.60 gm/mol), 5.6 mg trisodium citrate dihydrate, and 0.4 mg citric acid monohydrate was added 1.0 mg inositol, 1.5 mg gentisic acid, and 1.0 ml of 0.02 M HCl containing 5.40 mCi of "IIIIn. The pH of the solution was The solution was kept at room temperature and monitored for the amount of "III DTPA-octreotide using reverse-phase HPLC and a Beckman 170 radiometric detector. The HPLC method used a Hamilton PRP-1 column, 25 cm x 4.1 mm, 10 microns and a gradient system, linearly ramping from 100% A (A=10:90 ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) to 100% B (B=50:50 ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) over 60 minutes. The flow rate was 1.5 ml/min. The retention time of $^{\rm III}$ In DTPAoctreotide was 44-45 minutes.

The purity of the ^{III}In labeled peptide was 94% immediately post-reconstitution and 94% at 48 hours post-reconstitution.

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Example 5

Preparation of "In labeled DTPA-octreotide with gentisic acid additive.

To a vial containing 10 μ g DTPA-octreotide (N-[3,6,9,9-tetrakis(carboxymethyl)-3,6,9-triazanonanoyl]-D-phenylalanyl-L-hemicystyl-L-phenylalanyl-D-tryptophyl-L-lysyl-L-threonyl-L-hemicystyl-L-threoninol cyclic (2+7) disulfide, MW=1394.60 gm/mol) in 10 μ l of water was added 1 ml of a degassed stock solution of 2.3 mg gentisic acid and 41.4 mg sodium gentisate dihydrate and 25 μ l of 0.02 M HCl containing 4.66 mCi of ¹¹¹In. The pH of the solution was 4.2. The solution was kept at room temperature and monitored for the amount of ¹¹¹In DTPA-octreotide using reverse-phase HPLC and a Beckman 170 radiometric detector. The HPLC method used a Hamilton PRP-1 column, 25 cm x 4.1 mm, 10 microns, eluting with mobile phase A (A=20:10:70

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acetonitrile:ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) for 9 minutes, then eluting with mobile phase B (B=50:10:40 acetonitrile:ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) from 9 to 15 minutes. The flow rate was 1.2 ml/min. The retention time of ¹¹¹In DTPA-octreotide was 9-10 minutes.

The purity of the ¹¹¹In labeled peptide was 97% immediately post-reconstitution and 97% at 72 hours post-reconstitution.

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Example 6

Preparation of ""In labeled DTPA-octreotide with acetate and nicotinic acid additives.

To a lyophilized kit containing 10 μ g DTPA-octrectide (N-[3,6,9,9-tetrakis(carboxymethyl)-3,6,9-triazanonanoyl]-15 D-phenylalanyl-L-hemicystyl-L-phenylalanyl-D-tryptophyl-Llysyl-L-threonyl-L-hemicystyl-L-threoninol cyclic (2-7) disulfide, MW=1394.60 gm/mol) and acetate buffer, pH 4, was added 12.3 mg nicotinic acid and 1.0 ml of 0.02 M HCl containing 2.45 mCi of "IIIn. The pH of the solution was 20 The solution was kept at room temperature and 3.8. monitored for the amount of "III DTPA-octreotide using reverse-phase HPLC and a Beckman 170 radiometric detector. The HPLC method used a Hamilton PRP-1 column, 25 cm x 4.1 mm, 10 microns and a gradient system, linearly ramping from 25 100% A (A=10:90 ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) to 100% B (B=50:50 ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) over 60 minutes. The flow rate was 1.5 ml/min. The retention time of "In DTPAoctreotide was 44-45 minutes. 30

The purity of the ""In labeled peptide was 90% immediately post-reconstitution and 80% at 24 hours post-reconstitution.

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Example 7

Preparation of "In labeled DTPA-octreotide with citrate, inositol, and resorcinol additives.

To a lyophilized kit containing 10 μ g DTPA-octreotide (N-[3,6,9,9-tetrakis(carboxymethyl)-3,6,9-triazanonanoyl]-D-phenylalanyl-L-hemicystyl-L-phenylalanyl-D-tryptophyl-Llysyl-L-threonyl-L-hemicystyl-L-threoninol cyclic (2-7) disulfide, MW=1394.60 gm/mol), 5.6 mg trisodium citrate dihydrate, and 0.4 mg citric acid monohydrate was added 1.0 mg inositol, 11 mg resorcinol, and 1.0 ml of 0.02 M HCl 10 containing 5.17 mCi of "In. The pH of the solution was The solution was kept at room temperature and 4.5. monitored for the amount of "III DTPA-octreotide using reverse-phase HPLC and a Beckman 170 radiometric detector. The HPLC method used a Hamilton PRP-1 column, 25 cm x 4.1 15 mm, 10 microns and a gradient system, linearly ramping from 100% A (A=10:90 ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) to 43:57 A:B (B=50:50 ethanol:water, 10 mM tetrabutyl ammonium phosphate, pH 3) over 10 minutes, then held at 43:57 A:B for 15 minutes, then linearly ramped to 20 100% B over 10 minutes. The flow rate was 1.5 ml/min. retention time of "III DTPA-octreotide was 30 minutes.

The purity of the ¹¹¹In labeled peptide was 81% immediately post-reconstitution (also, 6% of an ¹¹¹In transfer ligand complex) and 87% at 120 hours post-reconstitution.

The results of Examples 1-7 are summarized in Table 1. These results demonstrate that gentisic acid, either alone or in combination with other stabilizers, is very effective at preventing radiolysis of radiolabeled peptides.

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	Example	Additive(s)	<u>Table 1</u> Initial Purity	Sta	ability
5	1	citrate	87%	50%	@ 15 hrs
10	2	citrate inositol	94%	73.5%	0 24 hrs
10	3	citrate inositol ascorbic acid	94%	86%	0 22 hrs
15	4	citrate inositol gentisic acid	94%	94%	@ 48 hrs
20	5	gentisic acid	97%	97%	0 72 hrs
20	6	acetate nicotinic acid	90%	80%	@ 24 hrs
25	7	inositol resorcinol	87%*	87%	@ 120 hrs

*81% "III DTPA-octreotide and 6% transfer ligand complex.

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Example 8 Preparation of 123 I labeled LH-RH.

This preparation is done in a well-ventilated fume hood equipped with a charcoal filter to absorb any volatile iodine.

In a 1.5 ml polyethylene tube, add 70 μ l of buffer A (A=150 mM sodium phosphate, pH 7.4), 10 μ l of a 0.1 mM solution of LH-RH (des-gly¹⁰,[D-ala⁶]-LH-RH ethylamide) in 0.1 M acetic acid, and 5 mCi of ¹²³I in 5 μ l of 0.01 M NaOH. The reaction is started by adding 10 μ l of a chloramine T solution (0.5 mg chloramine T/ml buffer A) to the peptide solution and is mixed by cautious aspiration and expulsion of the reaction mixture with the pipet used to add chloramine T. After 1 minute, 100 μ l of sodium metabisulfite solution (1 mg sodium metabisulfite/ml buffer A) is added to terminate the reaction and the reaction is

mixed with the pipet as explained previously. The reaction is kept in the hood a few minutes to allow any volatile iodine to exhaust from the reaction vial into the hood.

The solution is loaded onto a PD-10 column (G-25M Sephadex PD-10 column; Pharmacia LKB Biotechnology Inc.; Piscataway, NJ) eluting with PBS, pH 7.4, and 0.5 ml The 123 I labeled LH-RH is in fractions are collected. fractions 6 and 7, which are subsequently combined. of this solution is transferred to another vial containing 2 mg of gentisic acid.

The 123I LH-RH solutions are kept at room temperature and monitored for the amount of iodinated peptide and the amount of free iodide using reverse-phase HPLC and a Beckman 170 radiometric detector. The HPLC method uses a Hamilton PRP-3 column, 15 cm x 4.1 mm, 10 microns and a 100% Α eluting from system gradient 100% В (B=50:50to HCl) acetonitrile:water, 6 mM acetonitrile:water, 6 mM HCl) over 30 minutes. Free iodide elutes at the void volume in this system while the 123I LH-RH is retained and has a longer retention time. The solution containing the radiolabeled peptide without the gentisic evidence autoradiolytic of shows stabilizer degradation by the appearance of an increasing amount of free iodide in solution, as determined by HPLC analysis.

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Example 9

Preparation of 186Re labeled N3S-NR-Lu-10.

The NR-Lu-10 monoclonal antibody (NeoRx Corporation, seattle, WA) is labeled with 186Re using the pre-formed chelate approach developed by NeoRx and described in U.S. 30 Patent No. 4,965,392 and European Patent Publication Number The ligand used to form the 186Re complex is a 0284071. [N-(S-ethoxyethyl ester of tetrafluorophenyl mercapto)acetyl amino) adipoylglycyl]glycine. The 186Re complex formed is referred to as a \$^{186}\$Re-N_3S complex, since

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the 186 Re is coordinated by the three nitrogens and the one sulfur atom of the ligand. The 186 Re complex is formed by reducing 0.5 ml of 186 Re perrhenate (400 mCi, 4 mCi/ μ g Re) contained in a purged 10 ml vial with 0.5 ml of a deoxygenated stock solution of 1 mg stannous chloride, 10 mg gentisic acid, and 250 mg citric acid. The solution is kept at room temperature for 15 minutes. To this solution is added 0.2 ml of a 1 mg ligand/1 ml isopropanol stock solution. The solution is heated for 15 minutes at 80°C.

A solution of 25 mg of protein in 1.0 ml PBS is added to the vial containing the ¹⁸⁶Re-N₃S complex. The pH of the solution is adjusted to pH 9-9.5 with a 0.2 M sodium carbonate buffer. The solution is incubated at room temperature for 15 minutes. Given a typical conjugation yield of 50%, this reaction yields ¹⁸⁶Re-N₃S-NR-Lu-10 at a specific activity of 200 mCi ¹⁸⁶Re/25 mg NR-Lu-10.

The ¹⁸⁶Re-N₃S-NR-Lu-10 is purified using a PD-10 column (G-25M Sephadex PD-10 column; Pharmacia LKB Biotechnology Inc.; Piscataway, NJ) eluting with PBS, pH 7.4, collecting 0.5 ml fractions. The ¹⁸⁶Re-N₃S-NR-Lu-10 elutes in fractions 6 and 7, which are subsequently combined. Half of this solution is transferred to a vial containing 10 mg of gentisic acid.

The labeled protein is kept at room temperature and is monitored for the purity of ¹⁸⁶Re-N₃S-NR-Lu-10 using HPLC with a gel permeation column (Zorbax GF-250; 9.4 mm x 25.0 cm; DuPont, Wilmington, DE) eluting with 0.1 M phosphate, 0.1% sodium dodecyl sulfate, pH 6.7 mobile phase at a flow rate of 1 ml/min. Presence of autoradiolytic decomposition of the labeled monoclonal antibody is detected by the appearance of HPLC peaks that do not correspond to that for the ¹⁸⁶Re-N₃S-NR-Lu-10 monoclonal antibody. The radiolytic decomposition occurs in the solution that does not contain gentisic acid.

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The invention may be embodied in other specific forms without departing from its spirit or essential characteristics. The described embodiments are to be considered in all respects only as illustrative and not restrictive. The scope of the invention is, therefore, indicated by the appended claims rather than by the foregoing description. All changes which come within the meaning and range of equivalency of the claims are to be embraced within their scope.

10 What is claimed is:

- 1. A composition for preparing a stable radiolabeled peptide comprising:
 - a peptide capable of being labeled with a radionuclide; and
- a first stabilizer selected from the group consisting of gentisic acid, gentisyl alcohol and the water soluble salts, esters, derivatives, and mixtures thereof.
- 2. A composition as defined in claim 1, wherein the peptide is capable of being labeled with the radionuclide by use of a bifunctional chelate.
- 3. A composition as defined in claim 1, wherein the peptide is capable of being labeled with the radionuclide by direct labeling of the peptide.
- 4. A composition as defined in claim 1, wherein the peptide is capable of being labeled with the radionuclide 20 by covalent binding of the radionuclide to the peptide.
 - 5. A composition as defined in claim 1, further comprising a second stabilizer.
- 25 6. A composition as defined in claim 5, wherein the second stabilizer includes inositol.
 - 7. A composition as defined in claim 5, wherein the second stabilizer includes ascorbic acid.
 - 8. A composition as defined in claim 1, further comprising a buffer.
- A composition as defined in claim 1, wherein the
 composition is in the form of a lyophilized kit.

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- 10. A composition as defined in claim 1, wherein the peptide is labeled with a radionuclide.
- 11. A composition as defined in claim 10, wherein the5 composition is a liquid formulation.

- 12. A composition for preparing a stable radiolabeled protein comprising:
 - a protein capable of being labeled with a radionuclide; and
- a first stabilizer selected from the group consisting of gentisic acid, gentisyl alcohol and the water soluble salts, esters, derivatives, and mixtures thereof.
- 13. A composition as defined in claim 12, wherein the protein is capable of being labeled with the radionuclide by use of a bifunctional chelate.
- 14. A composition as defined in claim 12, wherein the 15 protein is capable of being labeled with the radionuclide by direct labeling of the peptide.
- 15. A composition as defined in claim 12, wherein the protein is capable of being labeled with the radionuclide20 by covalent binding of the radionuclide to the peptide.
 - 16. A composition as defined in claim 12, further comprising a second stabilizer.
- 25 17. A composition as defined in claim 16, wherein the second stabilizer includes inositol.
 - 18. A composition as defined in claim 16, wherein the second stabilizer includes ascorbic acid.
 - 19. A composition as defined in claim 12, further comprising a buffer.
- 20. A composition as defined in claim 12, wherein the 35 composition is in the form of a lyophilized kit.

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- 21. A composition as defined in claim 12, wherein the protein is labeled with a radionuclide.
- 22. A composition as defined in claim 21, wherein the5 composition is a liquid formulation.

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- 23. A composition for preparing a stable radiolabeled octreotide preparation comprising:
 - a quantity of DTPA-octreotide; and
- a stabilizer selected from the group consisting
 of gentisic acid, gentisyl alcohol, and the water
 soluble salts, esters, derivatives, and mixtures
 thereof.
- 24. A composition as defined in claim 23, further 10 comprising a second stabilizer.
 - 25. A composition as defined in claim 24, wherein the second stabilizer includes inositol.
- 26. A composition as defined in claim 24, wherein the second stabilizer includes ascorbic acid.
 - 27. A composition as defined in claim 24, further comprising a buffer.
 - 28. A composition as defined in claim 27, wherein the composition is in the form of a lyophilized kit.
- 29. A composition as defined in claim 27, wherein the 25 DTPA-octreotide is labeled with a ¹¹¹In.
 - 30. A composition as defined in claim 29, wherein the composition is a liquid formulation.

International Application No

I. CLASSIE	FICATION OF SUBJE	CT MATTER (if several classification syn	abols apply, indicate all) ⁶	
According	to International Patent	Classification (IPC) or to both National Cla	ssification and IPC	
Int.Cl	. 5 A61K49/0	2; A61K47/12		
II. FIELDS	SEARCHED			
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Classificat	ion System	C	lassification Symbols	
Int.Cl	. 5	A61K		
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		D TO BE RELEVANT ⁹		Relevant to Claim No.13
Category °	Citation of Do	cument, 11 with indication, where appropriat	e, of the relevant passages 12	Reevant to Claim No.23
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IV. CERTI	FICATION			
Date of the	Actual Completion of t	he International Search	Date of Mailing of this International Searc	h Report
	02 DECEME	BER 1992	2 5. 01. 93	
Internationa	Searching Authority		Signature of Authorized Officer	Julians
	EUROPE	AN PATENT OFFICE	DULLAART A.W.M. AW	Jullaans

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